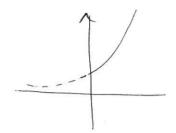
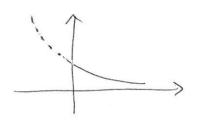
Applications

many real life problems are exponential models.



exponential growth



exponential decay

eg: population growth -> bacteria

- -> VIVUS
- > human population
- pop of species
- economic growth - compound interest

7

eg radio active decay
- heat transfer

→ Newtons law of
cooling

- atmospheric pressure (decreases exp with increasing neight above sea level)

model : y = Aekt

k = growth constant

For k>0 > exp growth

R<0 > exp decay

- 8. $P(t) = 15600e^{0.09t}$ describes the population of a city t years after 1984.
 - (a) What will the population be in 1994?
 - (b) How long will it take for the population to reach 100,000?

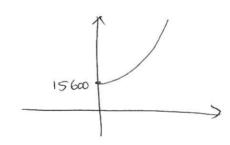
1

Make sure you understand the model.

- This is exponential growth since k=0.09 is pos



- Initially $\rightarrow t=0$: $P = 15600 e^{\circ} = 15600$ 80 in 1984 P = 15600



$$\frac{100000}{15600} = e^{0.09t}$$

$$\ln\left(\frac{1000}{156}\right) = \ln e^{0.09t}$$

:
$$0.09t = ln\left(\frac{1000}{156}\right)$$

$$t = \ln\left(\frac{1000}{156}\right)$$

9. The number of bacteria in a population, given by the formula $N(t) = N_0 e^{0.12t}$, has an initial population of 240000, where t is measured in hours. How long will it take for the population of bacteria to reach 250000?

7' Remember this ques from wk9.

N = number of bacteria t = time (hours)

We saw when t=0, N=240000

So the initial value No = 240 000

Our model looks like N(t) = 240 000 e

We want to know how long it will take for N=250000 ie: want t when N=250000

$$\frac{250000}{240000} = 240000 e^{0.12t}$$

$$\frac{250000}{240000} = e^{0.12t}$$

ie!
$$e^{0.12t} = \frac{25}{24}$$

$$\ln e^{0.12t} = \ln \left(\frac{25}{24} \right)$$

$$0.12t = \ln\left(\frac{25}{24}\right)$$

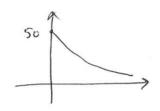
$$t = ln(\frac{2}{24})$$

= 0.341 hrs

1e! It will take 0.34 hr or 220 mins for the pop to reach 250000.

- 10. When a certain medical drug is administered to a patient, the number of milligrams remaining in the patient's bloodstream after t hours is modelled by $D(t) = 50e^{-0.2t}$.
 - (a) How many milligrams in the initial dose?
 - (b) How may milligrams will remain in the bloodstream after 3 hours?
 - (c) The patient needs to take a second dose of the drug once there is less than 5mg in their blood-stream. How many hours later does the second dose need to be administered?

$$D(t) = 50e^{-0.2t}$$



$$D(3) = 50 e^{-0.2(3)}$$

= 27.4 mg

$$50e^{-0.2t} = 5$$

$$e^{-0.2t} = \frac{5}{50}$$

$$-0.2t = ln(0.1)$$

$$t = \ln(0.1)$$

- Second dose should be administered after 11.51 hrs

1. The number of a certain species of frog is modelled by the function $N(t)=85e^{0.18t}$ where t is measured in years.

- (a) What is the initial population of frogs?
- (b) What will the population be after 3 years?
- (c) After how many years will the number of frogs reach 600?

a) initial population
$$\rightarrow$$
 Let $t=0$: $N=85e^{\circ}=85$

b) Want N when
$$t = 3$$
: $N = 85e^{0.18(3)}$
= 145.86

: After 3 yrs the population is 145

c) After how many years will population reach 600 ie: want t when N=600

$$85e^{0.18t} = 600$$

$$e^{0.18t} = \frac{600}{85}$$

$$0.18t = \ln(\frac{600}{85})$$

$$t = \ln(\frac{600}{85})$$

· Population reaches 600 in 10.86 years.

- 12. A hot bowl of soup is served at a dinner party. It starts to cool according to Newton's Law of Cooling so that its temperature at time t is given by $T(t)=18+62e^{-0.05t}$ where t is measured in minutes and T is measured in degrees Celsius(C).
 - (a) What is the initial temperature of the soup?
 - (b) What is the temperature after 10 minutes?
 - (c) After how long will the temperature be 37 degrees C?
 - (d) Make a graph of the temperature as a function of time.

$$T(t) = 18 + 62e^{-0.05t}$$
 $T = temp (°C)$
 $t = time (mins)$

a) Initial temp
$$\rightarrow$$
 want T when $t=0$
ie: $T=18+62e^\circ=18+62=80^\circ\text{C}$

b) temp ofter 10 mins
$$\rightarrow$$
 want T when $t = 10$
 $T(10) = 18 + 62e^{-0.05(t0)}$
= 55.6 °C

c) Now want time when
$$Temp=37$$
 1e: Want t when $T=37$

$$18+62e^{-0.05t}$$
 = 37
 $62e^{-0.05t}$ = 19
 $e^{-0.05t}$ = $\frac{19}{62}$

$$:=0.05t=\ln\left(\frac{19}{62}\right)$$

$$t = \ln\left(\frac{19}{62}\right)$$

. It takes 23.65 mins for the temp to drop to 37°C

so T= 62e

Half Life

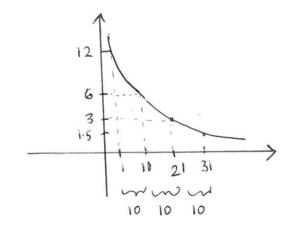
- Remember for exponential decay

The halving time is constant

Time it talces to halve in value

1
Half Life

eg :



Every 10 secs, the y-value halves Here the half life = 10 secs.

- This concept is very useful for certain exponential decay models.

hosphorous has a half life of 14 days. Suppose a sample of this substance has a mass of 300 mg.

- (a) Find the initial value and growth constant for this exponential decay function.
- (b) Hence write down the exponential decay function which models the amount of Phosphorous that remains in the sample after t days.

We are told Phosphorus has a half life of 14 days

This is telling us we have an exp (decay) model.

So were expect model to be of the form $P = P_0 e^{kt} \qquad (+ expect k + 0 be neg)$ P = amt of phos (mg) t = time(days)

We are told the sample has man of 300 mg ie. when t=0, P=300 & Initial value

-- Po = 300

so P=300ekt

Need to find k: We know half life is 14 days ie: Mass drops to half its value in 14 days. ie: when P=150, t=14

150 = 300 e 14k

 $\frac{150}{300} = e^{14k}$

1= e14k

 $-14k = ln(\frac{1}{2})$

K=ln(1/2)

= -0.0495...

This model has equation P(t) = 300 e

Notice: I didn't need the initial amount to find the growth constant.

If the half life is 14 days then when t=14, $P=\frac{P_0}{2}$

So
$$\frac{\rho_0}{2} = \rho_0 e^{i4k}$$

$$\frac{1}{2} = e^{i4k}$$
etc

- 14. The half life of cesium-137 is 30 years. Suppose we have a 100 g sample.
 - (a) Find a function that models the mass remaining after t years.
 - (b) How much of the sample will remain after 4000 years?
 - (c) After how long will only 18 g of the sample remain?

a) Let Cesium be modelled by the equation
$$C = C_0 e^{kt}$$
 [$c = cesium(g)$]. Initially sample = $100g$. $c_0 = 100$

· half life = 30 years
$$\rightarrow$$
 when $t = 30$, $c = \frac{C_0}{2}$
 $\frac{C_0}{2} = C_0 e^{30k}$

$$\frac{2}{e^{30k}} = \frac{1}{2}$$

$$30k = ln\left(\frac{1}{2}\right)$$

$$k = \ln(\frac{1}{2}) = -0.0231...$$

b) Want amount remaining after 4000 yrs

ie: Want C when t=4000

$$C = 100e^{-0.0231(4000)}$$

c) Want time when 18g remains

ie: Want t when C=18

$$100e^{-0.0231t} = 18$$

$$e^{-0.0231t} = \frac{18}{100} = \frac{9}{50}$$

$$-0.0231t = ln\left(\frac{9}{50}\right)$$

$$t = \ln \left(\frac{9}{50}\right) = 74.22$$

: It will take 74 yrs.

15. The half life of strontium-90 is 28 years. How long will it take a 50 mg sample to decay to a mass of 32 mg?

Strontium has half life of 28 yrs. [f=amt(mg)]

Let eqn be
$$P = P_0 e^{Kt}$$

Half life = 28 yrs \rightarrow when $t = 28$, $P = \frac{P_0}{2}$

Finding $K : \frac{P_0}{2} = P_0 e^{28K}$
 $\frac{1}{2} = e^{28K}$
 $= 28K$
 $K = \frac{\ln(\frac{1}{2})}{28} = -0.0247...$

initially sample = 50 mg
$$\Rightarrow$$
 Po = 50
so P = 50 e^{-0.0247t}

= 18.03

-- It takes & 18. yrs.

- 16. A wooden artifact from an ancient tomb contains 65% of the carbon-14 that is present in living trees. The half-life of carbon-14 is 5730 years. How long ago was the artifact made?
- First well find an eqn to model the decay: Let $C=C_0e^{kt}$ half life = 5730 \rightarrow when t=5730 $C=\frac{C_0}{2}$

$$\frac{c_0}{2} = c_0 e^{5730 k}$$

$$\frac{1}{2} = e^{5730 k}$$

$$\frac{1}{2} = e^{5730 k}$$

$$\frac{1}{2} = e^{1720 k}$$

we want age of artifact we know it has 65% of original amount.

ie: Want t when C=0.65Co

$$-0.00012t$$

$$0.65 = 6$$

$$0.65 = e^{-0.00012t}$$

$$0.65 = e^{-0.00012t}$$

$$1n(0.65) = -0.00012t$$

$$-10.00012$$

$$= 3561.13$$

The artifact was made 5 3561 yrs ago.